

Electrochemistry Multiple Choice Questions Answers

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Electrochemistry Multiple Choice Questions Answers

this guide covers electrochemistry multiple choice questions (mcqs) with answers and explanation.

Electrochemistry MCQs with Answers | Electrical Academia

Electrochemistry Practice Test with Answers. Question 1. The standard EMF for the given cell reaction $Zn + Cu^{2+} = Cu + Zn^{2+}$ is 1.10 V at 25 o C. The EMF for the cell reaction, when 0.1 M Cu^{2+} and 0.1 M Zn^{2+} solutions are used, at 25 o C is. Related: Question Bank for NEET Physics.

Electrochemistry Multiple Choice Questions Answers - JEE

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AP Chemistry: Electrochemistry Multiple Choice Answers 14. Questions 14-17 The spontaneous reaction that occurs when the

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cell in the picture operates is as follows: $2\text{Ag}^+ + \text{Cd (s)} \rightarrow 2\text{Ag (s)} + \text{Cd}^{2+}$ (A) Voltage increases. (B) Voltage decreases but remains $>$ zero.

AP Chemistry: Electrochemistry Multiple Choice Answers

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Electrochemistry Multiple Choice Questions Answers

Multiple Choice Questions 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____; the chemical change that occurs at this electrode is called _____.

Sample Questions - Chapter 21

APEF - Electrochemistry - Multiple Choice Questions A 1. Which of these chemical reactions is an oxidation-reduction reaction? A. $\text{Fe} + \text{S} \rightarrow \text{FeS}$ C. $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$ B. $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$ D. $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$ C 2. What happens to the oxidizing agent in an oxidation-reduction reaction? A. It is oxidized as it gains electrons. C.

APEF - Electrochemistry - Multiple Choice Questions ...

Multiple Choice Questions (Type-II) Note : In the following questions two or more than two options may be correct. The positive value of the standard electrode potential of Cu^{2+}/Cu indicates that _____. (i) this redox couple is a stronger reducing agent than the H^+/H_2 couple. (ii) this redox couple is a stronger oxidising agent than H^+/H_2 .

Class 12 Important Questions for Chemistry - Electrochemistry

Sample Questions - Chapter 21 Electrochemistry Examples of Multiple Choice Questions: 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____; the chemical change that occurs at this electrode is called _____. (a) anode, oxidation (b) anode, reduction (c) cathode, oxidation (d) cathode, reduction

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2. RESIST THE URGE

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NEET Chemistry Electrochemistry Questions Solved

31. During electrolysis of water the volume of oxygen liberated is 2.24 dm^3 . The volume of hydrogen liberated, under same conditions will be : 2.24 dm^3 . 1.12 dm^3 . 4.48 dm^3 . 0.56 dm^3 .
Answer. 32. In Down's method for the extraction of sodium, the melting point of the electrolyte is lowered by adding.

Important Questions of Electrochemistry for NEET Chemistry ...

Multiple choice questions. Try the following multiple choice questions to test your knowledge of this chapter. For each question there is one correct answer. The periodic table, physical constants and relative atomic masses needed for these problems are given on the inside covers of Chemistry, fourth edition by C.E. Housecroft and E.C. Constable. Once you have answered the questions, click on ...

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AP Chemistry: Electrochemistry Multiple-Choice Practice

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MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) The gain of electrons by an element is called _____. A)oxidation B)sublimation C)reduction D)disproportionation E)fractionation 2) _____ is reduced in the following reaction: $\text{Cr}_2\text{O}_7^{2-} + 6\text{S}_2\text{O}_3^{2-} + 14\text{H}^+ \rightarrow 2\text{Cr}^{3+} + 3\text{S}_4\text{O}_6^{2-} + 7\text{H}_2\text{O}$

A.P. Chemistry Practice Test - Ch. 17: Electrochemistry A

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NEET Chemistry Electrochemistry Multiple Choice Questions MCQs on Electrochemistry : 1. The aqueous solution of which of the following compounds is the best conductor of electric current?

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