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Lecture 34 . Kinetics I . Tutorial . 1) Equal numbers of moles of $\text{F}_2(\text{g})$ and $\text{ClO}_2(\text{g})$ are drawn into a vacuum where the following process takes place. $\text{F}_2(\text{g}) + 2 \text{ClO}_2(\text{g}) \rightleftharpoons 2 \text{FClO}_2(\text{g})$ a. At what time does the system reach equilibrium? The system reaches equilibrium about 45 min after the reactants are put in the container.

Lecture 34 Kinetics I Tutorial - AP Chemistry

Lecture 34 . Kinetics I . Worksheet . 1) The data below shows the change in concentration of dinitrogen pentoxide over time at 330 K, according to the following process.

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Lecture 34 Kinetics I Worksheet

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Lecture Slides. 34 pages. Lecture Slides. 30 pages. LECTURE NOTES. 4 pages. Lecture 16 ... Worksheet 3 on Kinetics Answer Key 1 Write the rate of the reaction $2O_3 \rightarrow 3O_2$ in terms of O_3 t 2 Consider the following stoichiometric reaction $2A + B \rightarrow C + D$ The reaction rates are measured with the following results Initial rate
1 2 1 A 0 2 4 2 B 0 1 1 ...

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